

MIXED-LIGAND COMPLEX OF COBALT(III) CONTAINING
 N,N'-BIS(2-PYRIDYLMETHYL)ETHYLENEDIAMINE AND ETHYLENEDIAMINE

Shunji UTSUNO , Arimichi HAYASHI, Susumu KONDO, and Minoru UTSUMI
 Department of Chemistry, Faculty of Science, Shizuoka University
 Oya, Shizuoka 422

The title complex was prepared and all the geometrical isomers and the diastereoisomers were separated. The structure of each isomer was assigned in terms of ^{13}C -NMR and CD spectra.

N,N'-Bis(2-pyridylmethyl)ethylenediamine ("penp") is a linear quadridentate ligand analogous to a well-known ligand of triethylenediamine ("trien"). Cobalt(III) complexes of penp were reported by Gibson et al.¹⁾ and Michelsen²⁾. Any trans-isomer has not been reported yet, probably because of its sterically hindered structure. Three isomers are possible for the complex of the type: $[\text{Co}(\text{XX})(\text{penp})]^{n+}$ (XX = two unidentate ligand in cis position or a bidentate ligand), as Fig. 1 shows. Two β -isomers are diastereoisomers. The nomenclature in the figure is based upon the corresponding structure of the trien complex³⁾. Only two isomers have been known in the case of $\text{X} = \text{Cl}^-$ ^{1,2)}.

The authors succeeded to prepare all the three isomers using ethylenediamine ("en") as a ligand, which occupied remaining coordination sites. The complex was prepared in the presence of active charcoal by stirring the mixture of en, $\text{en} \cdot 2\text{HCl}$, and $[\text{Co}(\text{OH})(\text{H}_2\text{O})(\text{penp})](\text{ClO}_4)_2 \cdot 2\text{H}_2\text{O}$, mole ratio of which was 1:4:2. The isomers were separated on a column of SP-sephadex C-25, using 0.25 M Na_2SO_4

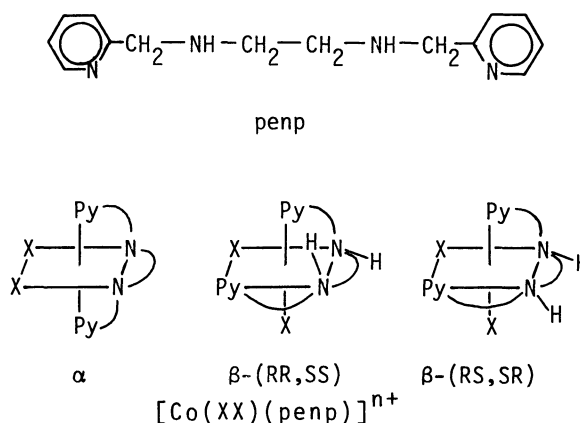


Fig. 1 Three isomers of penp complex.

acidified with H_2SO_4 (pH = 2.5) as an eluent. In this communication, three isomers are called in terms of their elution order as SH1-, SH2-, and SH3-en•penp. The SH2- and SH3-isomer could not be separated when neutral Na_2SO_4 or $\text{Na}_3\text{HP}_2\text{O}_7$ (pH = 6.8) was used as an eluent.

Table 1 The Analytical and Some Other Data of the Isomers

Complex perchlorate		Co(%)	H(%)	C(%)	N(%)	R*	$\nu_{\text{max}}, \text{cm}^{-1}(\epsilon)$
SH1-en•penp•H ₂ O	Found	8.75	4.08	28.13	12.24	1.00	21,600 (105)
	Calcd	8.70	4.16	28.36	12.40		29,700 (114)
SH2-en•penp•H ₂ O	Found	8.70	3.99	27.87	12.36	0.32	21,300 (177)
	Calcd	8.70	4.16	28.36	12.40		29,400 (139)
SH3-en•penp	Found	8.91	4.07	30.52	13.48	0.47	21,400 (172)
	Calcd	8.93	3.97	29.13	12.74		29,700 (160)

*Molar ratio of the yield to the yield of the SH1-isomer.

The solid complex was obtained as its perchlorate. Table 1 lists the elemental analyses, molar ratio of the yield, and wavenumbers and molar absorption coefficients of the d-d bands.

Of the three isomers, SH1-en•penp is assumed to be the α -isomer, because its molar absorption coefficients are significantly less than those of the others. This assumption is supported by the ^{13}C -NMR spectrum, which was recorded on a JNM FX-60 Fourier transform spectrometer, operating at 15.1 MHz. The two-fold axis, being present in the α -isomer, reduces the resonances to halves of all the carbon atoms.

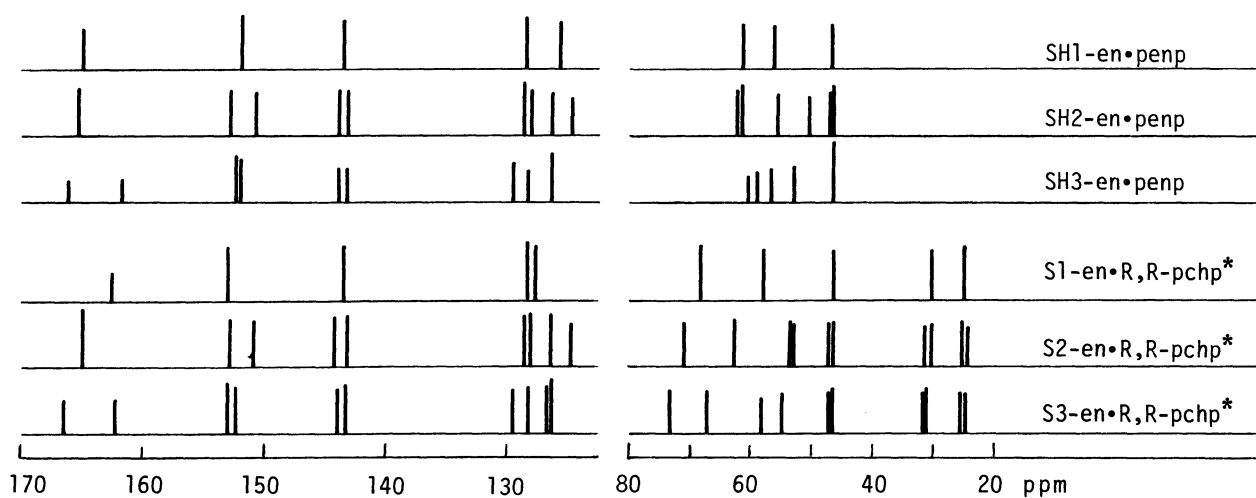


Fig. 2 Diagrammatic representation of the ^{13}C -NMR spectra in 0.05 M HCl with TMS as an external standard. *R,R-pchp is described in next page.

Two β -isomers may be characterized by their CD spectra. Fractional crystallization of β -[Co(en)(penp)][Co(cysu)₃]·nH₂O (cysu = L-cysteinesulphonate ion⁴); n is not determined), followed by separation of the isomers using a sephadex column with acidic Na₂SO₄ as an eluent, afforded each resolved isomer. Fig. 3 shows their CD spectra, which indicate that these isomers have Δ -configuration. The difference-curve is comparable with that of the reported complex, Δ - β -(RS)- and Δ - β -(RR)-[Co(gly)(trien)]I₂·H₂O³). Although the chromophore of both complexes is different, the resemblance of each difference-curve of both series of isomers is pronounced. Thus, SH2-en·penp and SH3-en·penp are tentatively assigned to the β -(RS,SR)- and β -(RR,SS)-isomer, respectively.

In order to ascertain their structures, N,N'-bis(2-pyridylmethyl)-1R,2R-cyclohexanediamine ("R,R-pchp") was prepared. When the ligand coordinates to a metal ion, central chelate ring is expected to be fixed to λ -conformation. Therefore, stereospecific formation of a Δ - β -(RS)-isomer and a Λ - β -(SS)-isomer is expected for the β -type isomers of [Co(en)(R,R-pchp)]³⁺. Actually, column chromatographic separation of the prepared complex with neutral Na₂SO₄ as an eluent yielded S1-, S2- and S3-en·R,R-pchp, which were found to be α -, β - and β -isomer, respectively, from

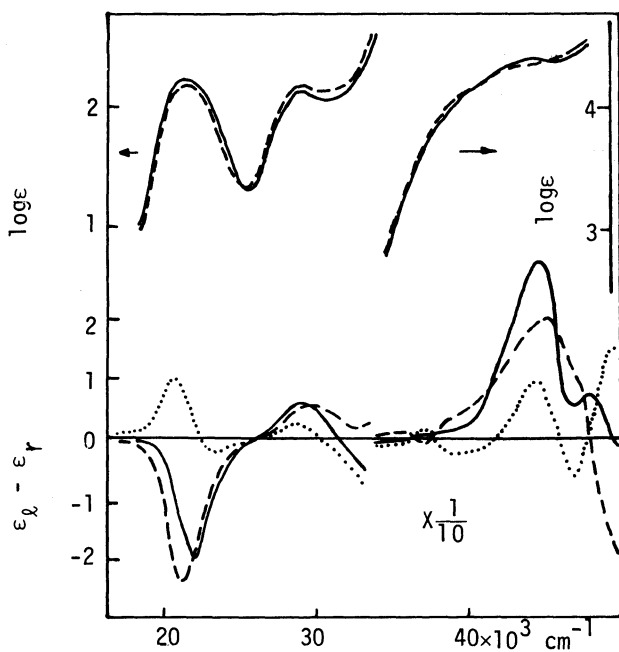


Fig. 3 AB and CD spectra of SH2-[Co(en)(penp)]³⁺ (—) and SH3-[Co(en)(penp)]³⁺ (----). Difference-curve (SH2 minus SH3) of both isomers (.....).

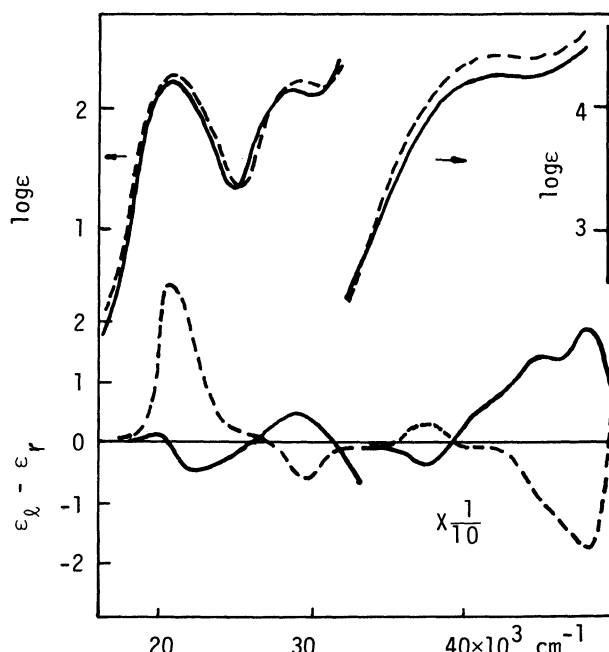


Fig. 4 AB and CD spectra of S2-[Co(en)(R,R-pchp)]³⁺ (—) and S3-[Co(en)(R,R-pchp)]³⁺ (----).

their ^{13}C -NMR spectra, which are shown in Fig. 2. The spectra also indicate that S2- and S3-en•R,R-pchp are conformationally equivalent to the SH2- and SH3-en•penp, respectively. On the other hand, the configuration of S3-en•R,R-pchp can be assigned as Λ by the CD spectrum shown in Fig. 4. The S2-isomer exhibits relatively weak spectrum in the region of the first absorption band, but its ultraviolet spectrum clearly indicate that it should have an opposite configuration to the S3-isomer, i. e., Δ -configuration. This means that the S2-isomer possesses β -(RS)-conformation and the S3-isomer possesses β -(SS)-conformation. This conclusion agrees well with the tentative assignment of the structure of SH2- and SH3-en•penp described above.

Acknowledgement. The authors wish to express their thanks to Prof. Junnosuke Fujita of Nagoya University for his valuable suggestions to this work and for his permission to use a Jasco J-40 CD recorder.

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(Received December 25, 1978)